

EVALUATION OF DIMENSIONAL STABILITY OF IMPRESSION MATERIALS IMMERSSED IN DISINFECTANT SOLUTIONS USING A METAL TRAY

AVALIAÇÃO DA ALTERAÇÃO DIMENSIONAL DE MATERIAIS DE MOLDAGEM IMERSOS EM SOLUÇÕES DESINFETANTES USANDO UMA MATRIZ METÁLICA

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ABSTRACT

Introduction: When impressions are taken, saliva and blood are removed from the material, and washing them does not always guarantee that the material is completely clean. Therefore, methods for disinfecting impressions (immersing in solutions) are a necessity, but they can affect the accuracy of dental impressions. This study was designed to evaluate the dimensional stability of dental impression materials immersed in disinfectant solutions. **Material and methods:** This study used 45 impressions of each of the following materials Impregun F® (polyether) and Hydrogun® (irreversible hydrocolloid). Sodium hypochlorite and glutaraldehyde were selected as disinfectants and the immersion times were 10 and 15 minutes. Impressions of each material were immersed in both solutions: 2% glutaraldehyde solution (Milton 1%) and sodium hypochlorite solution (Milton 1%), for 10 and 15 minutes. Impressions for 15 min. The control group without immersion in disinfectant solutions. **Results:** Neither group showed any statistically significant difference (ANOVA) in the measurements after being soaked in the two disinfectant solutions. For the 10-minute group, impressions were disinfected by sodium hypochlorite for 15 minutes. **Conclusion:** The dimensional stability of impressions during immersion practices for disinfection was not affected, except when sodium hypochlorite was used for 15 minutes.

UNITERMS: dimensional stability, dental impressions, impression materials.

SUMO

Introdução: Quando se tomam impressões de modelos para confecção de próteses, principalmente que são tomadas em áreas retentivas da boca, resíduos de saliva e sangue permanecem aderidos ao material, e durante a lavagem com água corrente, parte do sangue e microorganismos aderidos à superfície do molde são removidos. Assim torna-se necessária a desinfecção dos moldes para a qual os materiais de moldagem são submetidos a várias soluções desinfetantes. Este estudo teve como objetivo avaliar a estabilidade dimensional dos materiais de moldagem após a

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imersão em soluções desinfetantes de hipoclorito de sódio 1% e glutaraldeído 2%. **Material e métodos:** Foram realizadas 45 moldagens para cada um dos materiais: poliéter (Impregun F®), polissulfetos (Permlastic®), e hidrocolóide irreversível (Hydrogun®). 40 moldes de cada material foram imersos nas soluções desinfetantes variando-se o tempo entre 10 e 15 minutos. 10 moldes foram imersos em glutaraldeído a 2% por 10 min, 10 por 15 min, outras 10 imersos em hipoclorito de sódio a 1% por 10 min, e 10 por 15 min. Os demais moldes não foram imersos em nenhum desinfetante e serviram como grupo controle. **Resultados:** Nos moldes obtidos foram medidas as distâncias (AB, CD, AC, BD) em um microscópio digital Mitutoyo (TM 500) e os resultados foram submetidos à análise de variância a 2 critérios e as diferenças pelo teste de Tukey com intervalo de confiança de 95%. A análise dos resultados mostrou que não houve diferenças estatísticas significantes para os moldes de poliéter e polissulfeto quando comparados ao grupo controle. No entanto, para os moldes de alginato desinfetados em hipoclorito de sódio a 1% por 15 min, observou-se uma distorção estatisticamente significativa quando comparado ao grupo controle. **Conclusão:** Dentro dos limites dessa pesquisa pode-se concluir que a desinfecção dos moldes nas soluções de hipoclorito de sódio 1% e glutaraldeído 2% é uma prática segura, exceto para alginato que quando imerso durante 15 minutos em hipoclorito de sódio apresentou distorção, mas o mesmo não foi observado para esse material quando o tempo de 10 minutos foi usado.

UNITERMOS: estabilidade dimensional; desinfecção; materiais de r

INTRODUCTION

During the past few decades, authors have perpetuated the notion that dental impressions can lead to transmitting diseases, such as B hepatitis, tuberculosis, herpes and AIDS (Bond et al.³, 1983). When impressions are taken in retentive edentulous areas and subgingival preparations, blood has frequently been seen in the material, and washing alone does not clear it away, so there is no guarantee whatever that all organisms from the mouth that may have adhered to the impression surface have been removed (Look et al.¹⁷, Rios et al.¹⁸, 1996). Thus, disinfection of impressions have become a necessity; but these procedures affect the accuracy of dental impressions (Langervy et al.¹⁵, 1984; Setcos et al.²⁰, 1985; Setcos et al.²¹, 1985; Johnson et al.⁹, 1998; Dreyer et al.¹⁹, 1985, The American Dental Association published guidelines for disinfection in the dental office and laboratory. Contaminated impressions should be cleaned and handled in the dental laboratory (Panza et al.⁷, 1989).

Immersion and spray disinfectants, as well as many other solutions have been tested and proved to be effective for this purpose. The most reliable disinfection method is to immerse the impression to ensure that the disinfectant solution comes into contact with all the impression material surfaces and the tray (ADA⁵, 1977; Durr et al.⁸, 1987; Johnson et al.⁹, 1998; Johnson et al.¹⁰, 1998;

Langervy et al.¹⁵, 1984) Merchant et al.¹⁵, 1984)

It has been shown that chemical disinfection is effective against the hepatitis B virus after 10 minutes of exposure (Bond et al.³, 1983; Bond et al.²³, 1988). However other authors have tested different immersion times for dental impressions and found results similar to those of Bond and Tullner (Bergman et al.²²; Johnson et al.¹⁰, 1998). Thus, this study was conducted to evaluate the dimensional stability of impressions made with polyether, polysulfide and irreversible hydrocolloid impressions when immersed in two disinfectant solutions 2% glutaraldehyde (Glutacid® 2%) and sodium hypochlorite (Milton 1%) for periods of 10 and 15 minutes.

MATERIAL AND METHODS

A metal master model was used as described in The American Dental Association specification number 19 shown in Figure 1.

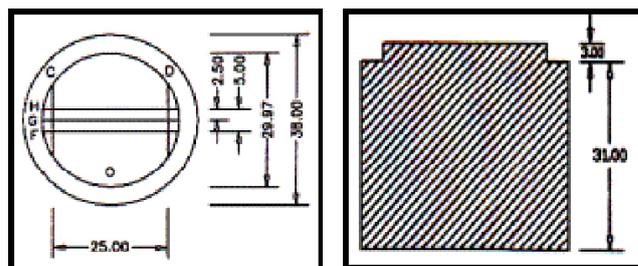
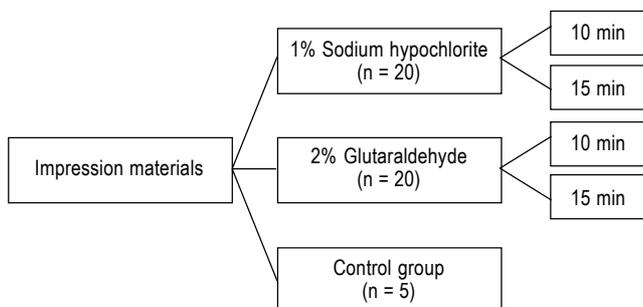


Figura 1 – Measurements of ADA specification n° 19.

Forty-five impressions were made with each of the following materials: Impregun F® (polyether), Permlastic® (polysulfide) and Hydrogun® (irreversible hydrocolloid). The selected disinfectants were: 1% sodium hypochlorite (Milton) and 2% glutaraldehyde (Glutaacid®). They were freshly prepared for each experiment.

The experimental groups were divided according to the diagram:



Ten impressions from each group of materials were immersed in glutaraldehyde solution for 10 min, and 10 impressions of the same materials were immersed for 15 min. The same procedure was carried out with the sodium hypochlorite solution. Five (5) impressions of each material were used as control group, without being immersed in any disinfectant solution.

The impression materials were dispersed and mixed according to the manufacturer's recommendations at room temperature (23°C). A gradual, constantly increasing pressure was applied to a perforated metal cast in order to avoid excess material. Afterwards, a 1 mm thick sheet was placed over the cast and the impression

was separated from the tray after 6 minutes. They were measured immediately after the impression procedure, to prevent any risk of distortion. Next, every impression was immersed in a disinfectant solution for 10 or 15 minutes, removed and rinsed under cold running water for 30 seconds and dried with compressed air. Immediately after drying, the original impressions were magnified × 30 and the distances between the lines AB, CD, AC and BD as shown in Figure 1 were measured three times by two examiners, for each elastomeric material. Irreversible hydrocolloid impressions were measured twice, because this material is more susceptible to syneresis and shrinkage (Andir et al.², 2002). Mitutoyo digital measuring microscope (TM500) with a sensitivity of 1.0 μm was used.

The analysis of variance (ANOVA) model was used to test the hypothesis that there was no difference between the disinfectants and the control, and between the 10 and 15 min treatments as used to analyze dimensional stability of each impression material after disinfection. The hypotheses tested were considered to be of confidence.

Forty-five impressions were made and divided into three groups: control and experimental groups. The graphic presentation of the data in Table 1 shows the means and standard deviations. The standard deviations ranged from 0.009 to 0.140 for alginate impressions, 0.004 to 0.092 for polyether and 0.016 to 0.149 for polysulfide. The results of the measurements obtained are expressed in millimeters.

TABLE 1 – means and standard deviations.

Material/Control	Means/AB	Means/CD	Means/AC	Means/BD
Alginate/control	24,062 ± 0.035	24,108 ± 0.053	4,197 ± 0.009	4,094 ± 0.016
Alginate/Hypochlorite 10 min	24,008 ± 0.046	24,077 ± 0.140	4,208 ± 0.029	4,108 ± 0.044
Alginate/Hypochlorite 15 min	23,871 ± 0.123	23,804 ± 0.122	4,191 ± 0.035	4,182 ± 0.095
Alginate/Glutaraldehyde 10 min	23,945 ± 0.126	23,935 ± 0.094	4,165 ± 0.047	4,079 ± 0.049
Alginate/Glutaraldehyde 15 min	23,983 ± 0.090	23,982 ± 0.098	4,166 ± 0.040	4,071 ± 0.024
Polyether/Control	23,997 ± 0.092	24,049 ± 0.056	4,182 ± 0.052	4,125 ± 0.045
Polyether/Hypochlorite 10 min	24,061 ± 0.021	24,079 ± 0.029	4,204 ± 0.007	4,139 ± 0.004
Polyether/Hypochlorite 15 min	24,083 ± 0.014	24,106 ± 0.012	4,215 ± 0.010	4,140 ± 0.006
Polyether/Glutaraldehyde 10 min	24,081 ± 0.008	24,079 ± 0.015	4,207 ± 0.015	4,123 ± 0.013
Polyether/Glutaraldehyde 15 min	24,089 ± 0.014	24,097 ± 0.015	4,208 ± 0.015	4,115 ± 0.010
Polysulfide/Control	24,103 ± 0.018	24,092 ± 0.019	4,220 ± 0.029	4,175 ± 0.049
Polysulfide/Hypochlorite 10 min	23,974 ± 0.149	23,998 ± 0.117	4,174 ± 0.016	4,141 ± 0.045
Polysulfide/Hypochlorite 15 min	24,062 ± 0.067	24,067 ± 0.021	4,171 ± 0.041	4,172 ± 0.024
Polysulfide/Glutaraldehyde 10 min	24,102 ± 0.039	24,080 ± 0.018	4,193 ± 0.034	4,154 ± 0.035
Polysulfide/Glutaraldehyde 15 min	23,994 ± 0.044	23,985 ± 0.107	4,188 ± 0.054	4,161 ± 0.031

The Tukey Test was used to analyze the results and showed no significant dimensional changes in polyether and polysulfide impressions in all periods of disinfection with glutaraldehyde and sodium hypochlorite ($p = 0.05$).

Neither polyether nor polysulfide impressions showed any statistically significant differences from their control measurements after soaking in the two disinfectant solutions. The null hypothesis for these experiments was that the mean distances measured in the control group were the same, irrespective of the impression being soaked in either one of the disinfectant solutions. However, after the alginate impressions were disinfected with sodium hypochlorite for 15 minutes a significant change was observed when compared with control group. A reduction in measurement represented alginate shrinkage. Macro and microscopic porosities were also found. The differences attained 0.3 mm (3×10^{-4} m) mainly in the largest distances (AB and CD).

DISCUSSION

The effects of different disinfectant solutions and times on three impression materials have been evaluated. There have been disagreements regarding their dimensional stability after the process (Drenon et al.⁷, 1989; Johnson et al.⁹, 1998; Setcos et al.²⁰, 1985; Thouati et al.²², 1997). According to the specifications provided by the United States Disease Control Centers, chemical disinfectants such as chlorine compounds, formaldehyde, glutaraldehydes, phenols, and iodophors have the potential to eliminate hepatitis, herpesviruses, and HIV viruses in 10 to 30 minutes (Matyas et al.¹⁴, 1990). In this study, the choice was to evaluate polysulfide and irreversible hydrocolloid impressions because they are hygroscopic and tend to shrink. Polyether and polyvinyl siloxanes. Sodium hypochlorite and glutaraldehyde were chosen as disinfectants because they are more widely used. Various types of tests were used in all arch casts, cavities for impressions. In this study, the American Dental Association standardization nº 19 was chosen for standardizing disinfection and impression procedures, because its usefulness in dentistry has been professionally recognized. Individual acrylic resin trays were not manufactured for the impressions because of the risk of water absorption and introduction of other variations. Stainless steel trays were used instead. For the same reason, the gypsum material was not poured.

The problem of disinfecting dental impressions (Lepe et al.¹², 2002), particularly irreversible hydrocolloid and hydrophilic ones, such as polyethers is a major concern. Herrera et al.⁷ (1986); Merchant et al.¹⁵ (1984), Langerwalter et al.¹¹ (1990) and Matyas et al.¹⁴ (1990) did not find dimensional change after the use of a 0.5% sodium hypochlorite solution for 30 minutes. In this study, the concentration of sodium hypochlorite was 2 times greater, which could explain the differences in the results. However, Minagi et al.¹⁶ (1987) showed that immersion in 2% glutaraldehyde for 60 minutes for irreversible hydrocolloid impression materials did not jeopardize surface details. When it was immersed in 1% sodium hypochlorite for 15 minutes, however, it was possible to observe statistically significant differences in surface alterations, like little cracks and imperfections of dentures. These differences would be transferred to the final results and subsequently to the fit of the dentures.

Although no significant linear dimensional changes were observed in polyether impressions for the different studies, other studies have shown that the use of this material adversely affects the accuracy of the final results of the casts. The same authors recommend that polyethers should not be disinfected with sodium hypochlorite, because they may lose surface details for periods exceeding 5 hours (Bergman et al.¹, 1974; Johnson et al.⁹, 1998; Drenon et al.⁷, 1989; Johnson et al.¹⁰, 1998; Setcos et al.²⁰, 1985; Owen et al.¹⁷, 1993; Sawyer et al.¹⁸, 1974). Disinfection of the polyether impressions using an alcoholic glutaraldehyde solution (2%) and sodium hypochlorite solution (1%), for periods of 10 and 15 minutes led to non-significant variations in all measured distances (AB, CD, AC, BD) without loss of accuracy or surface detail, a result consistent with previous studies. (Johnson et al.⁹, 1998; Drenon et al.⁷, 1989; Johnson et al.¹⁰, 1998). The results obtained with polysulfide impression disinfections shows no difference in mean values after all times of disinfection with both disinfectant solutions.

The results of this research are important to the dentist to select the appropriate disinfectant solution for specific clinical conditions such as fabrication of study models, removable partial dentures or fixed partial dentures.

CONCLUSION

Within the limits of this study it can be concluded that disinfect immersion practices

did not influence the dimensional stability of impressions obtained, except when sodium hypochlorite and a 15-minute immersion time were used for disinfecting irreversible hydrocolloid impressions.

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